

Chapter 10 Chemical Quantities Quiz Answer

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Chapter 10 Chemical Quantities 243 Section Review Objectives • Convert the mass of a substance to the number of moles of a substance, and the number of moles of a substance to mass • Calculate the volume of a quantity of gas at STP Vocabulary • Avogadro's hypothesis • standard temperature and pressure (STP) • molar volume Key Equations • mass (grams) number of moles

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10. Look at Figure 10.16 and Table 10.3. Name three compounds that have an empirical formula of CH. 11. Fill in the labels on the diagram below. MICROSCOPIC INTERPRETATION MACROSCOPIC INTERPRETATION SO₃ molecule 1 mol SO₃ SO₃ composed of composed of S atom and sulfur atoms (10²³) oxygen atoms 3 atoms and CHAPTER 10,Chemical Quantities(continued)

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SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER (pages 287–296)

Unit 4 discusses measurement in science as well as measurements that are specific to Chemistry. Chapter 3 deals with general math topics including significant digits and scientific notation. Chapter 10 introduces the idea of the mole and the use of equivalences to the mole as conversion factors.

Objectives Vocabulary Key Equations

CHAPTER 10: Chemical Quantities BASICS: • The basic unit that is used to determine the amount of a chemical substance is called a mole • A mole(mol) of a substance is equivalent to 6.02×10^{23} particles of that substance • The mole was founded by a scientist named Avagadro, and he decided to use the

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REVIEW OF CHAPTER 10 CHEMICAL QUANTITIES QUIZ ANSWER

- Although large in size, and filled with breathtaking photographs, this book includes so much more than the regular "picture book". Robyn's thoughtful words make you feel as if you are traveling right along with her and her famous camels. The story is engaging and heart-wrenching; and the reader runs through the same emotions that Robyn feels at each leg of the journey, from the tragedy of loss to the jubilation of completion.Beautiful and introspective - and very highly recommended.

- We have to remember that Mika Waltari wrote his books in the 1940s, long before other writers such as Ken Follett began their careers in historic fiction. In fact, Waltari is one of those authors that made historic fiction a popular genre of literature. His books ("The roman", "The etruscan", "Secret of the Kingdom", among others) are all based on great and accurate historic research and a central character, greatly developed and inserted in his background. One other thing that is repeated along his books is the presence, in different levels of transparency, of Christian elements (yes, even if it's ancient Egypt, as in this case).The egyptian of the title is Sinuhe, an orphan raised by an elder couple. Sinuhe's adopted father is a physician on a poor neighborhood in Thebes, Egypt's capital at the time (circa 1300 b.C.). Sinuhe chooses the profession of his father, but the difference is that he becomes the Pharaoh's physician, immersed in the court's game of deception, rivalries and connections. Sinuhe's time is the time of Pharaoh Amenhotep IV, who tried to change egyptian religion from polytheism to monotheism. The cult of Aton - the only god - was not well accepted by the religious class nor by the population, and the self-renamed Akhenaton (former Amenhotep) was one of the most hated rulers of all times.Sinuhe is a believable and complete character, although developed in an old-fashioned way. In fact, the whole book, its writing style and its characters have an old aura about them. Some of the situations and dialogues are extreme, and sometimes don't ring true, but those are rare occasions. Sinuhe and his faithful servant Kaptah travel through Egypt, Syria, Crete and back to Egypt, collecting information, meeting kings, girls and enemies. Sinuhe's life is full of things happening all the time. Waltari tells Sinuhe's envolvent story through the book. He was able to blend true facts with fiction in an excellent way, stablishing himself as one of the "fathers" of historic fiction.If you've never read Mika Waltari, "The egyptian" is a great way to begin.Grade 9/10