

# Chapter 9 Review Stoichiometry Section 2 Work

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 Section CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS  
 Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H<sub>2</sub> according to the following equation: N<sub>2</sub>(g) + 3H<sub>2</sub>(g) → 2NH<sub>3</sub>(g)  
 Answer the following questions in the space provided. 1. \_\_\_\_\_ The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.  
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## **REVIEW OF CHAPTER 9 REVIEW STOICHIOMETRY SECTION 2 WORK**

- I had such high hopes for this book - nominated for all those prizes! If you're looking for gay fiction then this book is for you. It

should be sold with the porn magazines - the first 100 pages ramble on and on about gay sex - but never get to a storyline. I tried to endure. I kept hoping there was a story! I grew up in Britain and thought this sounded like a great perspective on the Iron Lady (Prime Minister Margaret Thatcher) and 80's Britain. But it's just about a guy trying to fit in with a bunch of snobbish upper class twits! Oh yes, and there's lots of gay sex.

- I liked this even more than the Time Master Trilogy; the characters have more personality, and the stakes are higher. A fantastic read!