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revived in the eighteenth century, it took another hundred years before significantChapter 4: The Structure of the AtomChapter 4: The Structure of the Atom. 86Chapter 4. What You'll Learn. You will identify the experiments that led to the development of the nuclear model of atomic structure. You will describe the structure of the atom and differentiate among the subatomic particles that comprise it.Chapter 4: The Structure of the AtomSection 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these threeChapter 4 Atomic Structure -

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tiny indivisible particles called atoms. 2.
Atoms of the same element are
identical. The atom of any one element
are different from those of any other
element. 3. Atoms of different elements
can physically mix together or can
chemically combine in simple whole

number ratios to form compounds.
4.Chapter 4 Atomic Structure Flashcards
| QuizletSection 4.1 Defining the Atom
The Greek philosopher
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Electrons Protons Neutrons.

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Nuclear Atom. One change to Dalton's

atomic theory is that . atoms are

divisible. into subatomic particles:

Electrons, protons, and neutronsChapter

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The Greek philosopher Democritus (460

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later scientific theory, but did not explain

chemical behavior, and was not based

on the scientific method - but just

philosophy Dalton's Atomic Theory

(experiment based!)Chapter 4 Atomic

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Section Section 4.2 Structure of the

Nuclear Atom One change to Dalton's

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Structure - Henry County School

DistrictChapter 4 Atomic Structure

Section 4 1 Studying AtomsChapter 4.2

Structure of the Nuclear Atom Subatomic

Particles What are three kinds of

subatomic particles? Much of Dalton's

atomic theory is accepted today. • One

important change, however, is that

atoms are now known to be divisible. •

They can be broken down into even smaller, more fundamental particles, called subatomic particles.

Section 4.1 Defining the Atom The Greek philosopher Democritus (460 B.C. - 370 B.C.) was among the first to suggest the existence of atoms (from the Greek word "atomos") He believed that atoms were indivisible and indestructible His ideas did agree with later scientific theory, but did not explain chemical behavior, and was not based on the scientific method - but just philosophy Dalton's Atomic Theory (experiment based!)

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All elements are composed of tiny indivisible particles called atoms. 2. Atoms of the same element are identical. The atom of any one element

are different from those of any other element. 3. Atoms of different elements can physically mix together or can chemically combine in simple whole number ratios to form compounds. 4.

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Chapter 4 "Atomic Structure"

Chapter 4 Atomic Structure Section 4.2 Structure of the Nuclear

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Section 4.1 - Studying Atoms
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Chapter 4 Atomic Structure Section 4 1 Studying Atoms

Section 4.2 Structure of the Nuclear Atom OBJECTIVES: Describe the structure of atoms, according to the Rutherford atomic model. 10. Section 4.2 Structure of the Nuclear Atom One change to

Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons Protons Neutrons. 11.

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Section 4.2 Structure of the Nuclear Atom. One change to Dalton's atomic theory is that . atoms are divisible. into subatomic particles: Electrons, protons, and neutrons

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Section 4.1 Defining the Atom
The Greek philosopher Democritus (460 B.C. - 370 B.C.) was among the first to suggest the existence of atoms (from the Greek word

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must contain at least one proton.

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 opment of the nuclear model of
 atomic structure. You will describe the
 struc- ture of the atom and differ-
 entiate among the subatomic particles that
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REVIEW OF CHAPTER 4 ATOMIC STRUCTURE SECTION 41 STUDYING ATOMS ANSWERS

- Candide is an ambitious book. It should be an example for all 'would-be' writers all over the world. It is not less than a frontal attack on the greatest philosopher of Voltaire's time, Leibniz, for whom the world he lived in was 'the best possible'. 'Dear Pangloss (= know everything), when you were hanged, dissected, cruelly beaten, did you still think that everything was for the best in this world?' 'I still hold my original opinion', replied Pangloss, 'since Leibniz cannot be wrong.' This eventful text running with dazzling speed is a masterful mockery of Leibniz's

philosophy with its 'causes and effects', 'sufficient reasons', '(non)contingent events', 'freedoms and necessities', '(pre-established) harmonies', 'souls and evils' and 'natural laws': 'You expect to eat a Jesuit today; nothing could be more just, for natural law teaches us to kill our neighbor. If we don't exercise the right to eat him, it's because we have other things to make a good meal of.' Voltaire is a fundamental pessimist: 'Men have always slaughtered each other; they have always been liars, traitors, ingrates and thieves, cowardly, envious, greedy, ambitious, bloodthirsty, slanderous, lecherous, fanatical, hypocritical and foolish.' His philosophical solution is a flight from this brutal reality: 'let's work without theorizing; it's the only way to make life bearable.' The

only thing left is 'cultivate our garden.' This is a cowardly, selfish non-solution, to use Voltaire's own terms. Closing one's eyes for the realities of this world should not be an option. But how did Voltaire cultivate his garden? He profited handsomely from the slave trade. He even agreed that a ship for slave transport was named after him! A not so magnificent example of gardening. However, this brilliant 'cooking' of a philosopher's key ideas is a must read for all lovers of world literature. It should be a challenge for all ambitious writers.

- Candide ranks among the best crafted satirical essays on the human condition ever written in the western world. Too bad it wasn't required reading when I was in high school.